

## Coastal thermal springs in a foreland setting: The Santa Cesarea Terme system (Italy)



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### ABSTRACT

Carbonate aquifers in foreland tectonic settings can host important thermal springs although located in areas commonly not characterized by regional high heat flow values. In these cases, when thermal springs are located close or along the coastlines the subaerial and/or submarine thermal springs constitute the outflow of marine groundwater, flowing through localized fractures and karstic rock-volumes. This is the case of springs occurring along the south-easternmost portion of the Apulia region (Southern Italy) where few sulphurous and warm waters (22–33 °C) outflow in partially submerged caves located along the shoreline, thus supplying the historical spas of Santa Cesarea Terme. Here, with the aim to define the origin of the thermal fluids and their deep path, we carried out the geo-structural survey of the area and detailed hydrogeological and geochemical analyses of the thermal spring fluids. In particular, the isotopes  $\delta^{18}\text{O}$ ,  $\delta\text{D}$ ,  $^{13}\text{C}$  in DIC,  $^{34}\text{S}_{\text{Sulphate}}$ ,  $^{34}\text{S}_{\text{Sulphide}}$ ,  $^{3}\text{He}/^{4}\text{He}$  ratio and  $^{13}\text{C}$  in  $\text{CO}_2$  were used to define the origin of the thermal water and the recharge mechanism of the geothermal system while the isotopes  $^{3}\text{H}$  and  $^{14}\text{C}$  were determined for estimating the age of the thermal waters, resulting in older than roughly twenty thousands years BP. The results indicate that the thermal springs are fed by marine water, having reached Santa Cesarea Terme through a localized fracture network. This affects the evaporitic and carbonatic rocks that characterize the substratum of the Adriatic Sea in the offshore.

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### 1. Introduction

Most of the high temperature (>150 °C) water resources (Muffler and Cataldi, 1978) are mostly associated with recent volcanic activity occurring near plate boundaries, in the surroundings of crustal and mantle hot spot anomalies, or in continental extensional tectonic environments where coeval magma intrusions are volumetrically significant, thus determining heat flow values (>100 mW/m<sup>2</sup>; Goldstein et al., 2011).

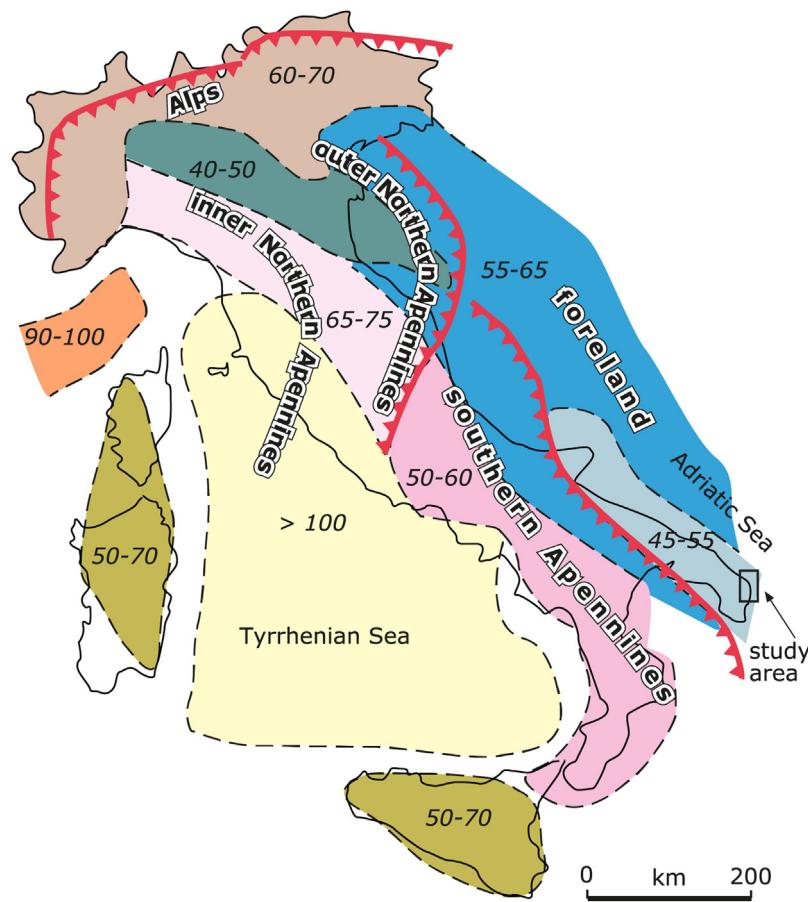
Thermal springs are some of the recognisable surface manifestations of thermal fluid circulation at depth. This can naturally occur in a volcanic context (e.g., hot springs on the island of Java in Purnomo and Pichler, 2014 and references therein) and/or in

areas where cooling magma is present only at mid-crustal depth (i.e. Larderello, Brogi et al., 2005). Obviously, thermal springs can be influenced by primary permeability differences among geological bodies, but very often, their location is driven by fault systems, defining permeable rock-volumes in rocks where the original texture would not permit a water storing. (i.e. Bürrchau springs in Rolker et al., 2015; Valdieri thermal springs in Baietto et al., 2008; southern Canadian Cordillera springs in Grasby and Hutcheon, 2001; Florida springs in Fanning et al., 1981).

As it concerns the Italian peninsula, thermal springs are connected to the original variability of permeability of the geological bodies and tectonic structures (e.g. Minissale, 2004). However this is a matter of fact that such a springs are mostly concentrated in the inner part of Apennines (Tyrrhenian sector; Fig. 1) where heat flow is high (>100 mW/m<sup>2</sup>) and it is characterized by reduced lithospheric and crustal thickness (about 50 and 25 km thick, respectively; Nicolich, 2001). Conversely, there are very few

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**Fig. 1.** Heat flow map of Italy: values of the conductive heat flow are given in  $\text{mW}/\text{m}^2$  (after Della Vedova et al., 2001; redrawn).

thermal springs are few in the eastern side of Italy, where heat flow decreases to 30–40  $\text{mW}/\text{m}^2$  (Adriatic sector; Della Vedova et al., 2001) and lithospheric and crustal thickness is about 120 and 45 km, respectively (Nicolich, 2001). These differences account for different tectonic environments that are extensional to the west, and compressional to the east, respectively (Mostardini and Merlini, 1986; Patacca and Scandone, 2007). Nevertheless, foreland areas, although characterized by low heat flow, are also affected by geothermal manifestations in spot-areas, with low temperature (about 25–28 °C) geothermal fluids at ground surface. This is the case of Santa Cesarea Terme thermal springs, located in the Apulia carbonate platform (Fig. 1), interpreted as the foreland (Ricchetti et al., 1988) of the southern Apennines collisional belt (Cretaceous-Pleistocene).

This paper focuses on the geothermal system associated with the Santa Cesarea springs, flowing out along a narrow coastal sector of the Salento peninsula (Figs. 1 and 2a) part of the Apulia carbonate platform, representing the southern Apennines foreland. As well known, seawater intrusion occurs along this coastal sector (Maggiore and Pagliarulo, 2004; Romanazzi et al., 2015). The Santa Cesarea spring water, with temperature ranging from 24 °C to 30 °C, is rich in sulphur (Calò and Tinelli, 1995) and nowadays is employed in a spa.

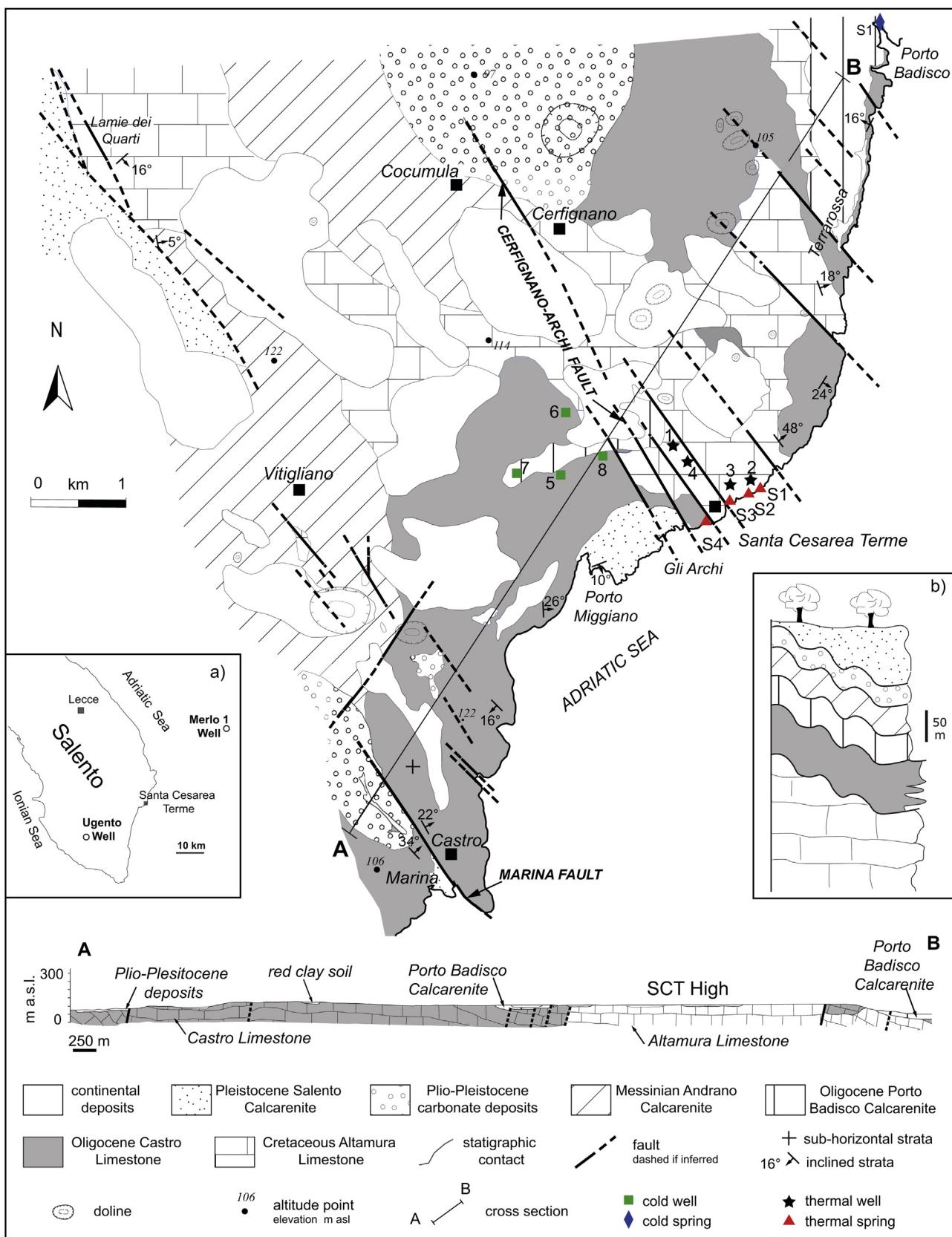
Several hypotheses on the origin of this warm groundwater were already proposed from volcanic (Milani, 1815) to a connate origin, associated to nearby hydrocarbon resources, flowing through fault zones dislocating the Dinaric foredeep (Maggiore and Pagliarulo, 2004). On the other hand, the sulphur content in the spring water might result from the sulphurization of deep groundwater, due to the activity of sulphate-reducing bacteria (Biginelli,

1899) affecting either the organic matter at depth or the coal included within Miocene outcrops (Zezza, 1980).

In order to contribute to the understanding of the Santa Cesarea Terme low-temperature geothermal system (referred as SCT system hereafter), we conducted a multidisciplinary research integrating the results from structural geology, hydrogeology, and hydrogeochemistry fields. In particular, the structural analyses were aimed at defining the structural channels, which controlled the circulation of the SCT thermal waters, whereas the groundwater flow and recharge were studied by hydrogeological in-situ investigations. Finally, the origin of the geothermal fluids and the role of water–rock interaction processes were argued from the hydrogeochemical analyses.

## 2. Geological and hydrogeological setting

The SCT system is hosted by the southern Apulian foreland (Ricchetti et al., 1988; Fig. 1). This is made up, from the bottom to the top, of: (a) pre-Cambrian crystalline basement postulated from geophysical data (Finetti, 2005 and references herein); (b) a Permian-Triassic succession, represented by siliciclastic and evaporitic (anhydrite and dolostone) sediments respectively, thicker than 3 km as encountered in an oil well located around 250 km to the North-West of Santa Cesarea Terme village (Puglia 1 well; ViDEPI Project, 2009), in particular the evaporite level is 400 m thick at least as indicated in the boreholes (Ricchetti et al., 1988); (c) Jurassic-Cretaceous carbonate succession, whose thickness surely overpasses 4.5 km (Ricchetti et al., 1988; Ugento 1 well in Fig. 2; ViDEPI Project, 2009); (d) Oligocene to Pleistocene carbonate to clastic sediments, extensively outcropping inland.



**Fig. 2.** Geological map of the study area with a representative geological section; (a) sketch of the Salento Peninsula with the location of the oil wells mentioned in the text; (b) stratigraphic log.

With reference to the studied area, the structural deeper outcropping geological body is the 600 m thick Altamura Limestone Formation (Late Cretaceous; Ricchetti et al., 1988), made up of well-bedded limestone, dolomitic limestone and dolostone (Fig. 2). This is overlain unconformably by the Late Oligocene Castro coral reef limestone Formation (Pomar et al., 2014 and references herein), as shown in Fig. 2. Both these formations are unconformably covered (Fig. 2) by poorly bedded neritic calciclastic deposits, known as the Oligocene Porto Badisco Calcarenite and the Messinian Andrano Calcarenite Formations (Bosellini et al., 1999; Pomar et al., 2014). Finally, cyclical Plio-Pleistocene shallow-water carbonate deposits rest at the top of the previously described successions (Fig. 2). Among these, the Pleistocene Salento Calcarenite Formation (Bossio et al., 2005 and references herein) deposited in a tectonically active slope environment, as testified by slumps and synsedimentary structures (Fig. 3).

All carbonate formations show intense karstification, both at the surface and at depth. Here caves and karsified and fissured horizons are interbedded with more compact limestone strata (Pepe and Parise, 2013; Romanazzi et al., 2015). The Mesozoic-Oligocene carbonate succession is part of the wide unconfined aquifer (defined as limestone aquifer herein), recharged through rainfall infiltration (Romanazzi et al., 2015), which involves more than the whole Salento (Fig. 1).

At SCT, the aquifer groundwater (calcium-bicarbonate water in Cotecchia, 1977; Polemio et al., 2009), hereinafter called fresh groundwater, floats on intruding seawater. However, a transition zone exists between the pure fresh groundwater and the saline water due seawater intrusion: within the transition zone the water salinity rapidly increases to depth from less than 0.5 g/l as total of dissolved solids (as TDS hereafter) to about 40 g/l of TDS, representing the seawater salinity (Cotecchia, 1977; Cotecchia et al., 2005; Polemio et al., 2009). Below the bottom of this zone spatially variable as main function of potentiometric head of fresh groundwater but roughly located at 150 m below sea level, salinity does not change with depth (Cotecchia, 1977; Cotecchia et al., 2005).

Groundwater temperature of limestone aquifer, usually equal to 14.5–16 °C inland, slightly increases along the flow paths, up to about 19–20 °C close to the coast (Cotecchia, 1977; Cotecchia et al., 2005; Romanazzi et al., 2015).

All summarised peculiarities of fresh groundwater were confirmed for the SCT sector.

The structure of the SCT offshore substratum is known through interpretation of reflection seismic lines (Mocnik, 2008; Del Ben et al., 2015 and reference herein) and stratigraphic borehole log (Merlo 1 well, ViDEPI Project, 2009). The Mesozoic-Tertiary carbonate platform is affected by normal faults defining a pelagic domain with isolated platforms (South Apulian basin; Mocnik, 2008; Del Ben et al., 2015) where Messinian deposits (evaporite, clays with

organic matter and mudstone, about 30 m thick in Merlo 1 well at a structural high), and 500 m thick Plio-Quaternary marine clastic sediments. As in other Mediterranean basins, these Messinian deposits are thicker toward the basin depocentre (Roveri et al., 2014).

The tectonic activity of the geological structures defining the Apulian offshore is documented by historical (during 1743 in Favali et al., 1990) and recent earthquakes as it was in 1974 ( $M = 4.9$ ) and in 2008 ( $M = 2.9$ ); the epicentres were located 60–120 km offshore (INGV, 2006; Mocnik, 2008). All of them were solved by focal mechanism, considering a strike-slip component (INGV, 2006; Mocnik, 2008; Di Bucci et al., 2011 and references herein).

### 3. Methods and data collection

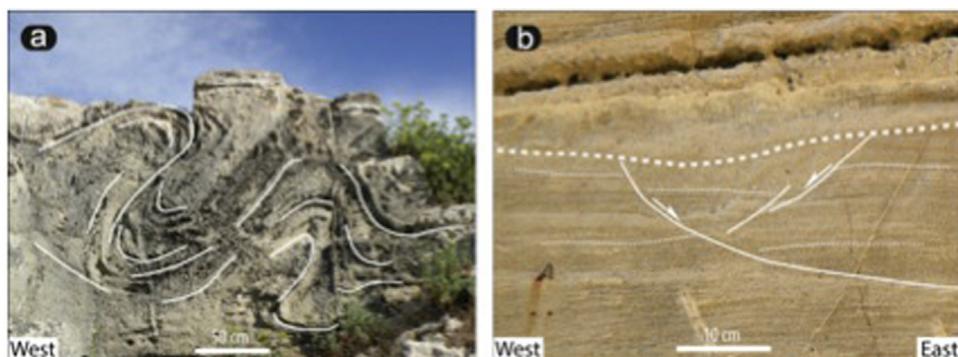
The geothermal exploration concerned about 50 km<sup>2</sup> (Fig. 2) an area including SCT and where fieldwork geological survey, hydrogeological field-testing and geochemical analyses were conducted.

The fieldwork was based on the classical approach of field mapping and structural geology, collecting structural and kinematic data in stations distributed in the study area on the basis of location of the main structures and suitable outcrops.

With regards to the hydrogeological analysis, sixteen stations for temperature measurements were selected. Considering thermal waters as those with temperature above 20 °C, therefore we selected 4 thermal and 2 cold springs, 4 thermal and 5 cold wells, seawater (Fig. 2).

Groundwater potentiometric head was measured within inland wells under no-pumping conditions (static head). From December 2011 to January 2012, groundwater temperature at Fetida and Gattulla springs (#S1, #S3 in Fig. 2) was measured hourly by means of probes connected to data loggers. The flow rate of these springs was acquired using an acoustic Doppler current profiler in their outflow section in the underground caves. Besides, electrical conductivity and water temperature were monitored at depth in both thermal wells n. 1 and n. 4 (Fig. 2) while the redox potential was measured in well n.1 only.

Groundwater samples for geochemical and isotopic analyses were collected both at spring sources and at wells (Tables 1–3). The water sampling was performed by standard sampling procedures, a part from well 3 where static sampling was used (two samples, at 45 m and 52 m below the ground surface; see also in Tables 1–2). Moreover, seven samples were taken at different depth during the drilling of well 4, after its flushing. All water samples were collected and stored in high-density polyethylene bottles (250 mL) with watertight caps. Samples for cation analysis were preserved by using an acidifying container (concentrated HNO<sub>3</sub> to a pH ≤ 2) while water samples for metals determination were firstly filtered

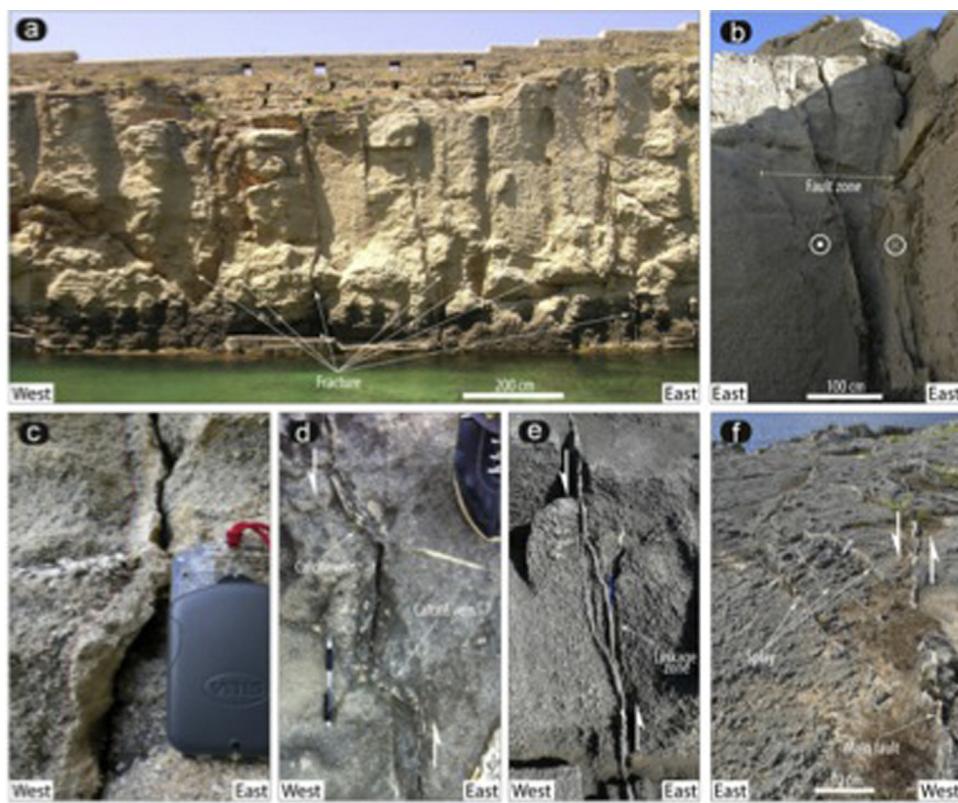


**Fig. 3.** Syn-sedimentary structures affecting the Quaternary calcarenite: (a) Highly non-cylindrical folds (slump) dissected by fractures; (b) Syn-sedimentary normal faults with centimetre to decimetre vertical offsets.

**Table 1**

Chemical and isotopic composition of water samples. Z) Altitude, WD) bottom well depth, EC) electric conductivity at 25 °C. n d = not determined.

|        | Sample  | Type-Name          | Z<br>(m a.s.l.) | WD<br>(m) | T<br>(°C) | E. C.<br>(mS/cm) | pH<br>(–) | Eh<br>(mV) | δD<br>(‰ SMOV) | δ <sup>18</sup> O<br>(‰ SMOV) | <sup>3</sup> H<br>(U.T.) | δ <sup>13</sup> C<br>PDB | <sup>14</sup> C<br>(BP) | <sup>14</sup> C<br>(pcm) | <sup>34</sup> S/ <sup>32</sup> S<br>(V-CDT) |
|--------|---------|--------------------|-----------------|-----------|-----------|------------------|-----------|------------|----------------|-------------------------------|--------------------------|--------------------------|-------------------------|--------------------------|---|
| Spring | S1      | Thermal-Fetida     | –               |           | 27.0      | 53.4             | 7.10      | -337       | 0.60           | 0.05                          | 1.03                     | -4.59                    | n d                     | n d                      | 19.7  |
|        | S2      | Thermal-Sulfurea   | –               |           | 21.1      | 59.3             | 7.69      | -85        | -13.8          | -2.74                         | 1.40                     | -1.41                    | n d                     | n d                      | 19.4  |
|        | S3      | Thermal-Gattulla   | –               |           | 24.5      | 41.4             | 6.20      | -265       | -0.40          | 0.37                          | 1.56                     | -8.35                    | n d                     | n d                      | 19.6  |
|        | S4      | Thermal-Solfatara  | –               |           | 25.7      | 40.1             | 7.54      | -220       | -6.70          | -1.47                         | n d                      | n d                      | n d                     | n d                      | n d   |
|        | S5      | Cold-Porto Badisco | –               |           | 21.0      | 7.60             | 7.80      | 169        | n d            | n d                           | n d                      | n d                      | n d                     | n d                      | n d   |
|        | S6      | Cold-Acquaviva     | –               |           | 17.8      | 27.2             | 7.53      | 210        | n d            | n d                           | n d                      | n d                      | n d                     | n d                      | n d   |
| Wells  | 1       | Thermal            | 115.4           | 205.0     | 27.5      | 62.1             | 6.88      | -263       | 5.9            | 1.78                          | 0.46                     | -1.55                    | 19849                   | 8.45                     | 20.0  |
|        | 2       | Thermal            | 33.57           | 50.0      | 32.7      | 64.2             | 6.71      | -268       | 5.9            | 1.57                          | 0.33                     | -1.95                    | 21050                   | 7.54                     | 19.9  |
|        | 3–45 m  | Thermal            | 41.76           | 200.0     | 21.4      | 7.09             | 7.57      | -237       | n d            | n d                           | n d                      | n d                      | n d                     | n d                      | n d   |
|        | 3–52 m  |                    | –               | –         | –         | –                | –         | -21.7      | -3.39          | n d                           | n d                      | n d                      | n d                     | n d                      | n d   |
|        | 4–125 m | Thermal (Vigor)    | 115.0           | 300.0     | 22.0      | 4.21             | 6.99      | -136       | n d            | n d                           | n d                      | n d                      | n d                     | n d                      | n d   |
|        | 4–151 m |                    | –               | –         | 24.0      | 52.5             | 7.80      | -399       | n d            | n d                           | n d                      | n d                      | n d                     | n d                      | n d   |
|        | 4–300 m |                    | –               | –         | 27.0      | 52.2             | 7.99      | -296       | n d            | n d                           | n d                      | n d                      | n d                     | n d                      | n d   |
|        | 5       | Cold               | 121.0           | –         | 19.2      | 0.44             | 8.70      | -21        | -34.2          | -5.47                         | 4.60                     | -12.81                   | n d                     | n d                      | n d   |
|        | 6       | Cold               | 117.0           | –         | 19.3      | 3.88             | 8.21      | 80         | -30.3          | -5.39                         | n d                      | n d                      | 4268                    | 58.5                     | n d   |
|        | 7       | Cold               | 123.0           | 130.0     | 20.1      | 2.29             | 8.27      | -39        | -30.2          | -5.57                         | n d                      | -12.83                   | n d                     | n d                      | n d   |
| Sea    | 8       | Cold               | 112.0           | –         | 19.8      | 0.95             | 8.08      | 74         | -33.3          | -5.64                         | n d                      | -13.48                   | n d                     | n d                      | n d   |
|        | 9       | Cold               | 108.6           | 116.4     | 16.0      | 0.87             | 7.05      | 75         | -33.2          | -5.78                         | 2.80                     | -10.58                   | n d                     | n d                      | n d   |
| Sea    | 10      | Sea Water          | –               | –         | 19.1      | 60.3             | 8.20      | 169        | 7.1            | 1.48                          | n d                      | n d                      | n d                     | n d                      | n d   |



**Fig. 4.** Examples of brittle structures affecting the Quaternary calcarenite; (a) m-thick spaced, sub-vertical, fractures cemented by calcite; (b) particular of a left-lateral oblique-slip fault zone characterized by anastomosed shear fractures cemented by calcite; (c) detail of the shear fracture cemented by calcite, supporting for infiltration of fluids; (d) extensional jogs filled by calcite in a left-lateral strike-slip fault zone; (e) anastomosed left-lateral strike-slip shear fractures cemented by calcite; (f) Splay system developed within the damage zone of a main left-lateral strike-slip fault.

through a cellulose acetate membrane (pore size 0.2 micrometres) and secondly, these were acidified by nitric acid.

Samples electrical conductivity (EC), pH, temperature (T), redox potential (Eh) and dissolved oxygen ( $O_2$ ) were measured directly in the field by means of a multiparametric probe. Chemical and isotopic compositions were produced at the CNR-IGG laboratories, following standard procedures by: (a) ion-chromatography for cations and anions (Table 2); (b) volumetric titration for  $HCO_3^-$  (Table 2); (c) inductively coupled plasma mass spectrometry for trace elements (Table 3) and isotopes (Table 1).

Isotope  $^{34}S$  (thermal water sampled at wells #2 and #4),  $^3H$  and  $^{18}O$ , were ascertained by the mass spectrometer at the Environmental Isotope Laboratory – Earth & Environmental Sciences Department (University of Waterloo, Canada). Finally, the isotope enrichment factor,  $\varepsilon$  equal to  $\delta^{34}S_{\text{Sulphate}} - \delta^{34}S_{\text{Sulphur}}$ , was defined following Wynn et al. (2010).

One stripped gas was obtained by the #2 thermal well by syringing 500 ml of water under vacuum, of which composition was obtained after gas-chromatographic analysis.

In the following,  $^{18}O/^{16}O$  and D/H ratio are expressed with the typical  $\delta$  notation referred to as V-SMOW, while the abundant ratio of  $^{34}S$  and  $^{32}S$  is generally given as  $\delta^{34}S$  relatively to VCDT. The overall precision of the analyses on major ions, is within 5%. Differently, the experimental errors are estimated in  $\pm 0.1\%$  for  $\delta^{13}C, \pm 1.0\%$  for  $\delta^2H, \pm 0.3\%$  for  $^{34}S$  (dissolved in both  $SO_4^{2-}$  and  $H_2S$ ) and  $\pm 0.5\%$  for  $^{18}O$  dissolved in  $SO_4^{2-}$ .

$^3H$  and  $^{14}C$  activities of the dissolved inorganic carbon in the water were used to constrain the age estimates of the study waters. The concentration of  $^3H$  was estimated by means of inductively coupled plasma mass spectrometry (Table 1). Conversely, the concentration of  $^{14}C$  for radiocarbon groundwater dating was achieved

by an accelerator mass spectrometry at the Diagnostic and Dating Centre of the Salento University (Italy). The results are shown in Table 1, expressed as both uncorrected BP apparent ages and percent modern carbon (pcm).

#### 4. Data analyses

##### 4.1. Structural data

The relationships between brittle geological structures and geothermal fluid flow were investigated through a structural and kinematic survey of the study area. The most significant structural and kinematic data are from the Salento Calcareous succession, the most recent sediment cropping out.

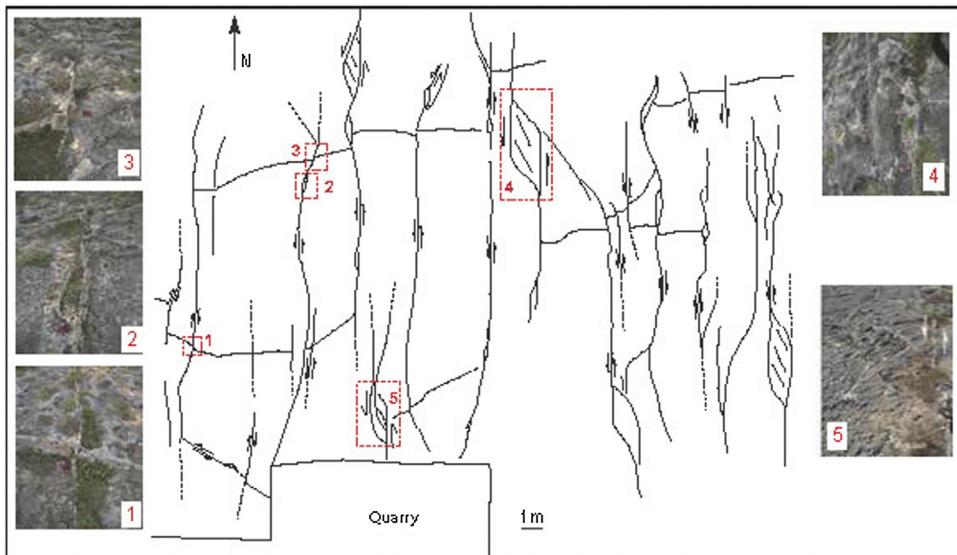
The deformation is characterized by subvertical, N-S trending faults (Fig. 4a), and associated minor fractures (Fig. 4b). The damage zone (Fig. 4b) is decimeter in thickness whereas the slip zone displays a cemented cataclasite (Fig. 4c). Fluids circulated through the cataclasite and extensional jogs providing calcite veins (Fig. 4d). No clear kinematic indicators on the fault surfaces have been recognized but the angular relationships between the main faults and associated fractures clearly suggest a left lateral strike-to-oblique slip movement (Fig. 4d,e and f). This kinematics is also confirmed by the detailed structural survey carried out in a key-area, as illustrated in Fig. 5, where the left lateral component is dominant.

Deformation in the calcareous substratum is defined by normal to left-lateral oblique slip faults. The most significant structure crops out in the Castro Harbor (Fig. 6) where one of this fault clearly dissects the carbonate succession and the overlying Quaternary deposits (Fig. 6a,c and d). The damage zone, about 8 m thick, is

**Table 2**

Chemical composition (in mg/L) of water samples. n d = not determined. n p = not present.

| Sample  | $\text{Li}^+$ | $\text{Na}^+$ | $\text{K}^+$ | $\text{NH}_4^+$ | $\text{Mg}^{2+}$ | $\text{Ca}^{2+}$ | $\text{Sr}^{2+}$ | $\text{F}^-$ | $\text{I}^-$ | $\text{Cl}^-$ | $\text{HCO}_3^-$ | $\text{Br}^-$ | $\text{NO}_3^-$ | $\text{SO}_4^{2-}$ | $\text{S}^{2-}$ |
|---------|---------------|---------------|--------------|-----------------|------------------|------------------|------------------|--------------|--------------|---------------|------------------|---------------|-----------------|--------------------|-----------------|
| S1      | 0.42          | 9219          | 347          | 1.41            | 959              | 908              | 12.7             | 118          | 0.29         | 17304         | 228.8            | 82.6          | 41.5            | 2878               | 3.12            |
| S2      | 0.05          | 12130         | 383          | 0.00            | 1358             | 784              | 10.4             | 9.14         | n d          | 20212         | 202.8            | 83.7          | 50.7            | 3322               | 0.61            |
| S3      | 0.27          | 5415          | 224          | 0.79            | 5766             | 574              | 7.59             | 76.4         | 0.28         | 9035          | 309.9            | 39.7          | 30.6            | 1833               | 2.51            |
| S4      | 0.34          | 10258         | 264          | 49.1            | 768              | 672              | 15.9             | 18.8         | n d          | 16429         | 244.0            | 76.3          | 0.00            | 1926               | n d             |
| S5      | 0.03          | 1351          | 72.8         | 0.95            | 164              | 165              | 1.47             | 10.9         | 4.40         | 2189          | 311.1            | 7.98          | 25.6            | 424                | n.p             |
| S6      | 0.05          | 2210          | 152          | 0.00            | 514              | 154              | 0.00             | 0.38         | 19.2         | 4538          | 299.0            | 15.2          | 8.98            | 802                | n.p             |
| 1       | 0.55          | 13477         | 509          | 0.00            | 1385             | 1311             | 18.7             | 36.1         | 0.33         | 23923         | 195.2            | 113           | 29.5            | 3813               | 21.7            |
| 2       | 0.45          | 13484         | 506          | 1.63            | 1339             | 1553             | 20.9             | 54.8         | 0.31         | 24369         | 187.0            | 123           | 62.9            | 3848               | 31.6            |
| 3–45 m  | 0.02          | 628           | 38.2         | 0.00            | 55.6             | 75.4             | 0.00             | 0.54         | 1.46         | 1109          | 354.0            | 5.83          | 0.9             | 128                | n.p             |
| 3–52 m  | 0.14          | 4630          | 245          | 0.00            | 685              | 485              | 0.01             | 0.60         | 0.27         | 9220          | 213.0            | 25.6          | 0.55            | 1167               | 7.50            |
| 4–125 m | 0.02          | 643           | 25.5         | 0.00            | 72.0             | 118              | 0.25             | 0.37         | n d          | 1260          | 311.0            | 4.00          | 22.0            | 156                | n.p             |
| 4–151 m | 0.25          | 12269         | 378          | 0.00            | 1005             | 492              | 13.0             | 1.60         | n d          | 22681         | 189.1            | 74.0          | 11.9            | 2640               | 0.33            |
| 4–300 m | 0.45          | 10180         | 438          | 0.00            | 991              | 1218             | 19.1             | 1.88         | n d          | 17858         | 213.5            | 77.0          | 0.00            | 2087               | 0.74            |
| 5       | 0.00          | 15.4          | 0.24         | 0.24            | 2.99             | 64.9             | 0.36             | 1.26         | n d          | 26.5          | 170.9            | 0.13          | 12.8            | 6.60               | n.p             |
| 6       | 0.01          | 544           | 26.3         | 4.29            | 63.2             | 132              | 0.00             | 2.58         | 0.28         | 1118          | 274.6            | 4.42          | 47.5            | 140                | n.p             |
| 7       | 0.01          | 319           | 11.4         | 0.00            | 38.8             | 102              | 0.26             | 2.55         | 3.53         | 645           | 268.5            | 2.51          | 7.72            | 83.0               | n.p             |
| 8       | 0.00          | 60.1          | 30.6         | 0.00            | 13.1             | 90.5             | 0.47             | 0.93         | 0.63         | 137           | 241.0            | 0.49          | 74.9            | 35.2               | n.p             |
| 9       | 0.00          | 66.5          | 36.9         | 5.84            | 32.2             | 110              | 0.34             | 6.38         | 0.57         | 138           | 392.8            | 0.32          | 23.1            | 67.5               | n.p             |
| 10      | 0.18          | 11617         | 417          | 0.00            | 1489             | 520              | 14.2             | 10.6         | n d          | 23371         | 189.1            | 94.2          | 0.00            | 2859               | n.p             |

**Fig. 5.** Scan lines resulted from structural survey performed at Porto Miggiano (locality in Fig. 2).

mostly localized in the fault hanging wall. It is defined by coalescing fractures with attitudes coherent with the fault geometry and kinematics and by a cataclastic breccia (Fig. 6d) made up of carbonate elements in calcareous cement. Kinematic indicators consist of grooves, mega-grooves and slickensides, visible on slip surfaces both on the main fault scarp (Fig. 6b and e) or on minor surfaces within the cataclasite (Fig. 6f).

The Castro Marina fault is therefore an example of the high-angle faults dissecting the carbonate rocks with a NW-SE trend in the surroundings of the study area (Fig. 2). Thus, in their arrangement, this fault system defines structural highs where the villages of Castro and Santa Cesarea are located (Fig. 2), referred as Castro High and SCT High in the following.

As regards the age of activity of this system, the Castro Marina fault, as well as the other companion structures, dissects the Salento Calcarenite succession as recognized in the geological map (Fig. 2). On the basis of this evidence and on the fact that this fault cross-cuts the Quaternary breccia, we interpret such a fault system as active since post-Pleistocene, at least. Therefore, in the period

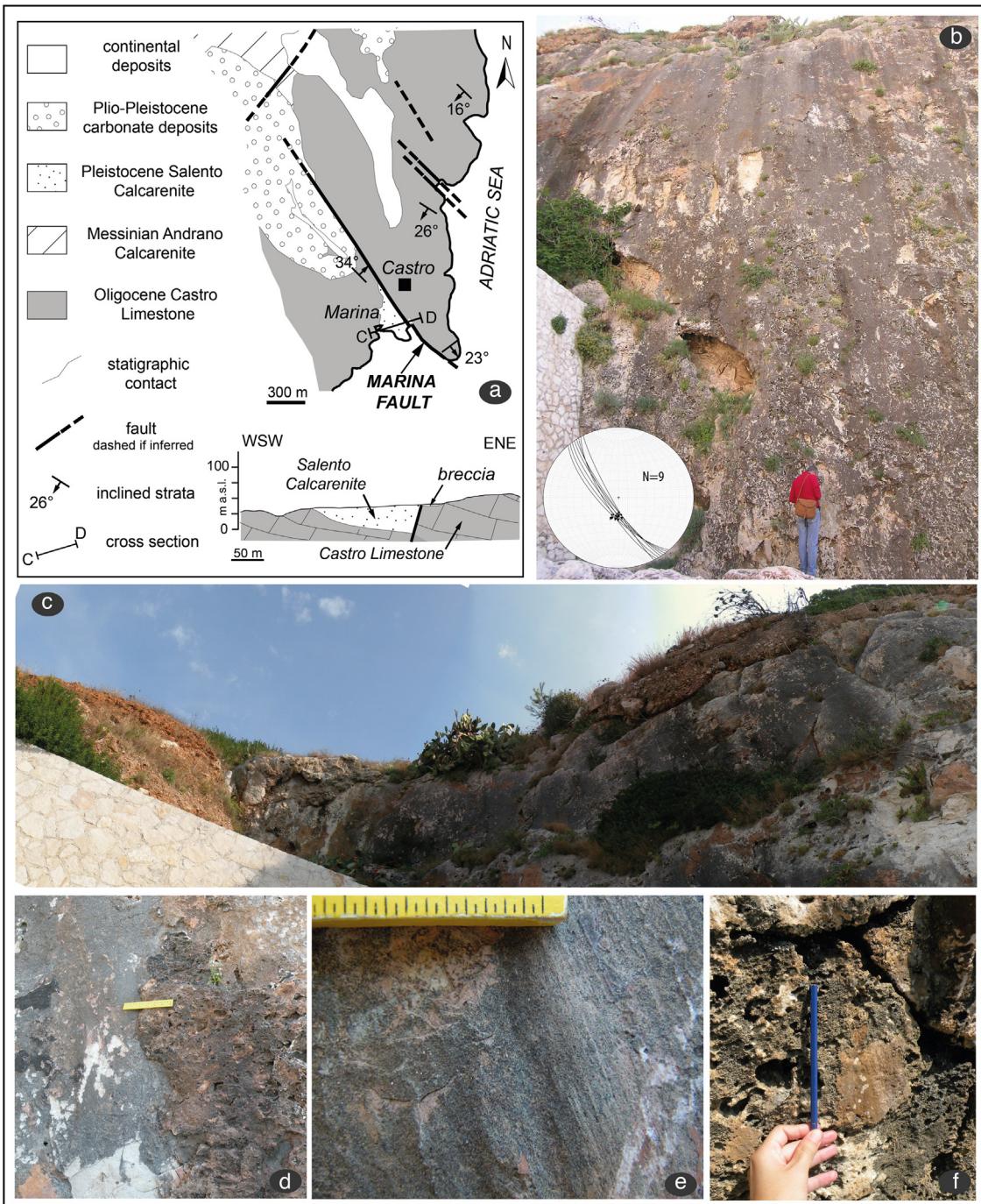
encompassed by Pleistocene and Present, the area is affected by two coeval fault systems, N-S and NW-SE trending, respectively.

Regarding the kinematic and geometric compatibility between the two different fault systems, some considerations could be proposed: (a) the occurrence of extensional jogs and linkage zones NW-SE oriented in the Salento Calcarenite succession (Fig. 4) suggests a possible explanation for a coherent deformational context in which N-S and NW-SE faults can be framed; (b) it is a matter of fact that, at map scale, all the N-S trending faults are not highlighted, although this can be a consequence of the recent fault activity, implying reduced offsets; (c) the NW-SE faults display a left-lateral oblique-slip movement and a significant offset, exceeding 10 m, implying a fault activity older than the N-S oriented fault system. It is therefore suggested that the Pleistocene-Present deformation determined the N-S fault trend and partially reactivated the already existing NW-SE trending faults. In this framework, a localized improvement of permeability, along almost vertical structural channels is envisaged.

Finally, the described fault systems had a strong influence on the development of the main karst landforms (doline, karst valley and

**Table 3**Chemical composition (in  $\mu\text{g/L}$ ) of water samples. –not determined.

| Sample | Ag    | Al   | As   | Au    | B     | Ba   | Be    | Bi    | Cd   | Co   | Cr   | Cs   | Cu    | Fe    | Hg  | Ho   | Mn   | Mo   | Ni   | Pb    | Rb   | Sb   | Se    | Si   | U    | V    | W    | Zn   |
|--------|-------|------|------|-------|-------|------|-------|-------|------|------|------|------|-------|-------|-----|------|------|------|------|-------|------|------|-------|------|------|------|------|------|
| S1     | <5    | <100 | 64.0 | <5    | 6057  | 50.0 | <5    | <5    | <2   | <50  | 1.00 | 24.0 | <1000 | <10   | <1  | <5   | <10  | <20  | 1740 | 111   | <5   | 158  | 4405  | <2   | 139  | <2   | <50  |      |
| S2     | <5    | <100 | 258  | <5    | 5208  | 40.0 | <5    | <5    | 5    | 6    | <50  | 2.00 | 11.0  | <1000 | <10 | <1   | 12   | <10  | <20  | 5722  | 116  | 6.00 | 206   | 4353 | <2   | 89.0 | <2   | 500  |
| S3     | <5    | <100 | <50  | <5    | 3327  | 37.0 | <5    | <5    | <2   | <50  | 1.00 | <10  | <1000 | <10   | <1  | 5    | <10  | <20  | 1132 | 71.0  | <5   | 111  | <4000 | <2   | 23.0 | <2   | <50  |      |
| S5     | <0.5  | <10  | 8.0  | <0.5  | 527   | 39.1 | <0.5  | <0.5  | <0.5 | <0.2 | <5   | <0.1 | 2.00  | <100  | <1  | <0.1 | <0.5 | 2    | <2   | 341   | 13.9 | <0.5 | 20.0  | 4067 | 1.6  | 16.0 | <0.2 | <5   |
| S6     | <0.5  | 2.87 | 14.5 | <0.5  | 1.56  | 28.0 | <0.5  | <0.5  | <0.5 | <0.2 | <5   | 0.20 | –     | 4.58  | <1  | <0.1 | 0.11 | –    | <2   | –     | 31.8 | 0.12 | –     | –    | –    | –    | <0.2 | <5   |
| 1      | <5    | <100 | 67.0 | <5    | 8906  | 51.0 | <5    | <5    | <2   | <50  | 2.00 | <10  | <1000 | <10   | <1  | 9.00 | <10  | <20  | 3375 | 159   | <5   | 226  | 4152  | <2   | 68.0 | <2   | <50  |      |
| 2      | <5    | <100 | 59.0 | <5    | 10590 | 57.0 | <5    | <5    | <2   | <50  | 2.00 | <10  | <1000 | <10   | <1  | 24.0 | <10  | 572  | 4137 | 165   | <5   | 270  | 4973  | <2   | <20  | <2   | <50  |      |
| 3–45 m | <5    | 4.89 | 2.14 | <5    | 1.66  | 24.3 | <5    | <5    | <2   | <50  | 0.15 | <10  | 115   | <10   | <1  | 37.6 | <10  | <20  | –    | 21.7  | 0.08 | –    | –     | <2   | <20  | <2   | <50  |      |
| 3–52 m | <5    | 3.70 | 16.4 | <5    | 3.06  | 31.0 | <5    | <5    | <5   | <2   | <50  | 0.40 | <10   | 88.6  | <10 | <1   | 25.0 | <10  | <20  | –     | 46.4 | 0.14 | –     | –    | <2   | <20  | <2   | <50  |
| 5      | <0.05 | 0.20 | 0.13 | <0.05 | 0.01  | 2.45 | <0.05 | <0.05 | –    | –    | <5   | 0.10 | –     | 35.4  | <1  | <0.1 | 26.1 | –    | –    | –     | 0.12 | 0.06 | –     | –    | –    | –    | <0.2 | –    |
| 6      | <0.05 | 3.44 | 1.53 | <0.05 | 0.18  | 7.47 | <0.05 | <0.05 | –    | –    | <5   | 0.10 | –     | 66.1  | <1  | <0.1 | 3.03 | –    | –    | –     | 4.00 | 0.04 | –     | –    | –    | –    | <0.2 | –    |
| 7      | <0.05 | 0.27 | 0.26 | <0.05 | 0.04  | 2.39 | <0.05 | <0.05 | –    | –    | <5   | 0.10 | –     | 354   | <1  | <0.1 | 44.6 | –    | –    | –     | 0.60 | 0.03 | –     | –    | –    | –    | <0.2 | –    |
| 8      | <0.05 | 6.23 | 0.80 | <0.05 | 0.04  | 3.44 | <0.05 | <0.05 | –    | –    | <5   | 0.10 | –     | 9.27  | <1  | <0.1 | 2.10 | –    | –    | –     | 14.4 | 0.05 | –     | –    | –    | –    | <0.2 | –    |
| 9      | <0.05 | 6.00 | 1.00 | <0.05 | 38.0  | 47.8 | <0.05 | <0.05 | 0.11 | 0.02 | 7.20 | 0.01 | 7.70  | 47.0  | <1  | <0.1 | 5.61 | 0.60 | 0.30 | 26.20 | 1.20 | 0.10 | 1.20  | 3815 | 1.92 | 4.10 | <0.2 | 10.7 |



**Fig. 6.** (a) Geological map of the Castro area with a representative geological section – (b) Main fault scarp exposed in the Castro harbour and relative stereographic diagram (equiareal diagram, lower hemisphere) indicating a main left-lateral oblique-slip kinematics; (c) Quaternary breccias dissected by the Castro Fault attesting a Quaternary age of the Castro Fault; (d) epigenetic calcite cementing the Castro Fault cataclasite indicating the infiltration of meteoric water within the fault zone; (e–f) kinematic indicators on the slip surface consist of mechanical striation (e) and calcite + Fe-hydroxides thin crusts (f).

so on) identified in the studied territory (Ardizzone et al., 2012). In particular, the cave growing directions (thermal and no thermal caves) resulted also to be deeply influenced by the described fault systems, thus indicating the structural control on the fluid flow.

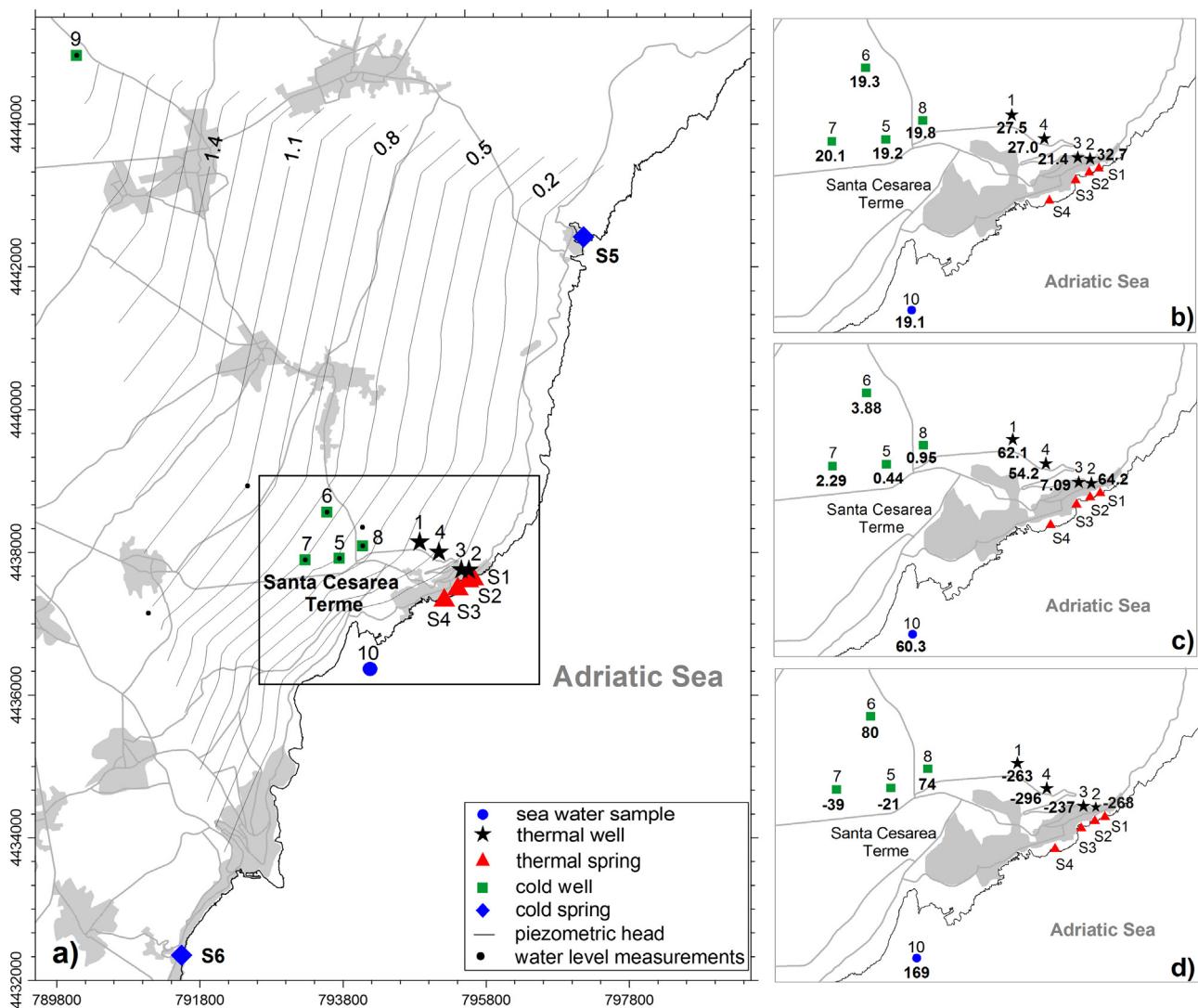
#### 4.2. Hydrogeological data

Fig. 7a shows the piezometric surface contour map for the limestone aquifer revealing a general groundwater flow direction towards southeast or seaward. This confirms the common trend characterizing the whole Salento region, where fresh and cold

groundwater of the limestone aquifer flows from inland recharge areas to the outflow coastal areas (Cotecchia et al., 2005). In this framework, the piezometric head changes from about 2 m a.s.l. to 0 m a.s.l. in SCT (Fig. 7a).

The limestone groundwater temperature changes from 16 °C (outside the study area, in the recharge zone) to about 20 °C (Fig. 7b); the electric conductivity is less than 3.6 mS/cm (Fig. 7c) while the total dissolved solids are generally less than 500 mg/L; the redox potential ranges close to zero (Fig. 7d).

Taking into consideration that the T, EC, and Eh, the higher absolute values (maps in Fig. 7b-d) were recorded in the wells where



**Fig. 7.** Piezometric surface map (limestone aquifer, m a.s.l.), well and spring code (Table 1) (a); Monitoring survey carried out in June 2012 and in June 2014 only for well 4: temperature map ( $^{\circ}\text{C}$ , b), electrical conductivity map (mS/cm at 25  $^{\circ}\text{C}$ , c), and Eh map (mV, d).

the thermal springs and wells are settled (Figs. 2 and 7). Here, in fact, groundwater temperatures ranges between 22 and 33  $^{\circ}\text{C}$  with a TDS up to 5 mg/L. Their related piezometric head is fairly higher than in the nearby cold wells (e.g. 1 m a.s.l. estimated for thermal well 4 respect to 0.6 m a.s.l. obtained for cold well 7; see Fig. 2).

The water temperature, the water level, and the outflow yield measured in the Fetida and Gattulla springs ensued to change during the time. As shown in Fig. 8a, Fetida spring displays a temperature ranging between 14.6  $^{\circ}\text{C}$  and 28.4  $^{\circ}\text{C}$  while its water level is from -0.726 m to 0.708 m, both of them respect to the sea level. On the other hand, the Gattulla spring is characterized by the water temperature from 13.6  $^{\circ}\text{C}$  to 29.7  $^{\circ}\text{C}$  and water level between -0.421 m s.l. and 0.864 m.s.l (Fig. 8b). The flow yield was measured two times resulting 0.283-0.312 m<sup>3</sup>/s and 0.0014-0.0097 m<sup>3</sup>/s for Fetida and Gattulla springs, respectively. These temporal changes are explained as effects of the tide fluctuations (Visintin, 1944; Calò and Tinelli, 1995) affecting the coastal sector under study.

Fig. 9 shows the multiparametric logs achieved for the thermal wells n. 1 and n.4 (Fig. 2), where water table is s. T and EC, measured for both wells, increase to depth while Eh decreases moving downwards in well 1 (Fig. 9). In this latter, abrupt changes were recorded for all the measured parameters almost at the same depth, of about

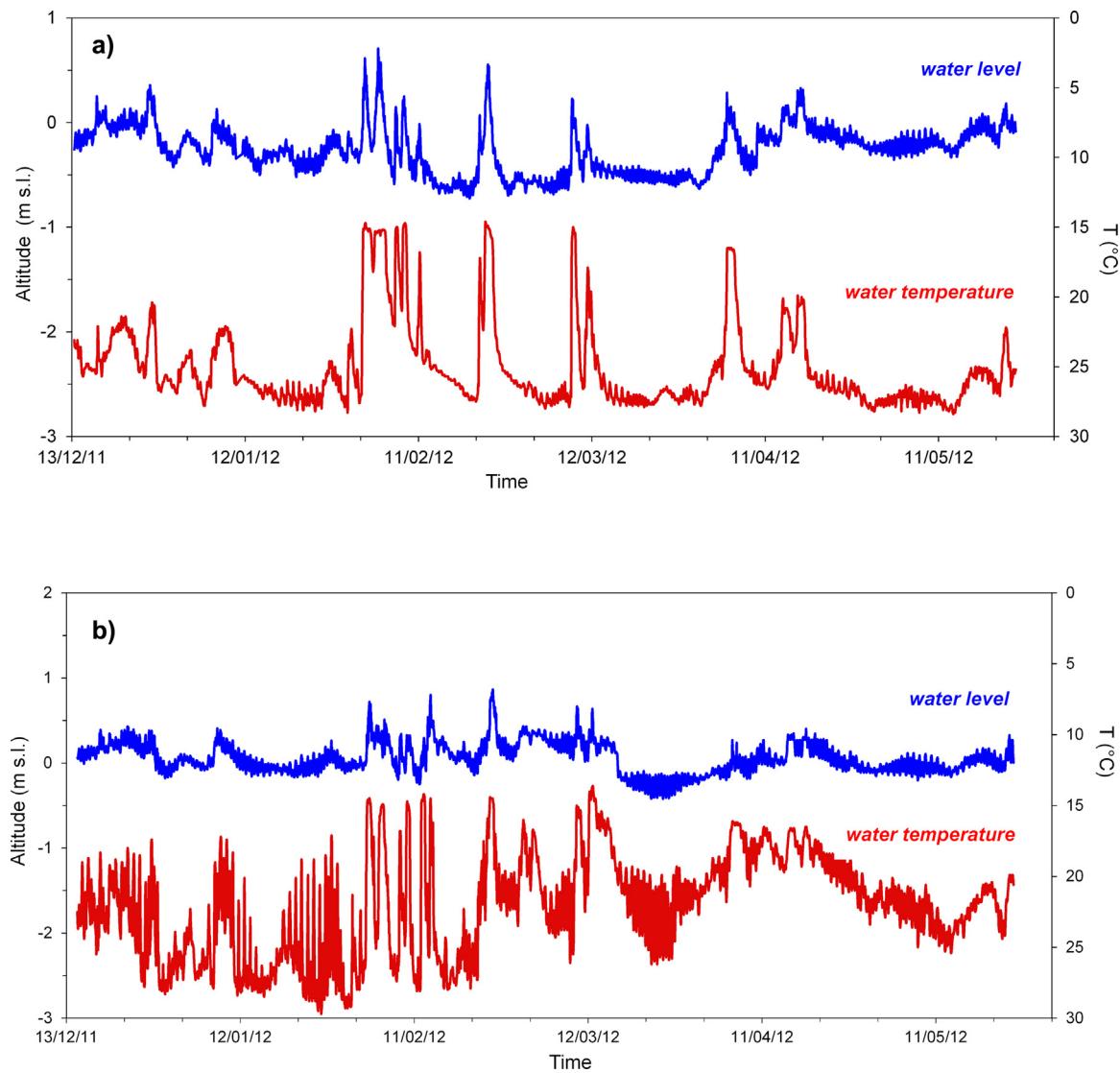
138 m b.g.l. (below ground level). However, after that, groundwater values resulted fairly constant: T shifted from 24  $^{\circ}\text{C}$  to 28.1  $^{\circ}\text{C}$ , EC changed from 10 mS/cm to 63 mS/cm, and Eh from almost 20 mV down to -400 mV (Fig. 9a). Contrariwise, EC changed suddenly with depth and T increased gradually from 19.5  $^{\circ}\text{C}$  to 27.4  $^{\circ}\text{C}$  (Fig. 9b). Moreover, EC reached its maximum (67 mS/cm) at about 295 m b.g.l. (Fig. 9b). Only to the temperature log of well 4 (Fig. 9b) shows a local increase, that can be explained as the consequence of hotter water plumes, coming from more fractured and karstified horizons intercepted at that depth.

#### 4.3. Geochemical data

Tables 1-3 report the chemical and the isotopic composition determined for all the water samples. Samples #1 and # 2 represent the thermal waters, sample #5 is indicative of the fresh groundwater and sample # 10 is the seawater; all the other samples have a variable geochemistry, as discussed later on.

The thermal waters could be classified as Na-Cl type while the fresh water as Ca-HCO<sub>3</sub> type (Fig. 10).

The thermal waters have the highest concentrations of Ca<sup>2+</sup> and the lowest content of Mg<sup>2+</sup> (Table 1); they are under-



**Fig. 8.** Variations in water temperature and water level measured in the Fetida (a) and Gattula (b) thermal springs.

saturated in calcite and dolomite ( $\text{SI}_{\text{calcium}} = -0.38/-0.50$  and  $\text{SI}_{\text{dolomite}} = -0.30/-0.63$ ) while they result nearly saturated with respect to gypsum (Fig. 11). The thermal waters have also the highest  $\text{Li}^+$ ,  $\text{Sr}^{2+}$ ,  $\text{B}^{3+}$  and  $\text{Br}^-$  contents (Tables 2–3). Taking 4.6 mg/L as reference value for the boron in seawater being almost constant throughout the world seas (White, 1957; Kharaka and Hanor, 2003), the  $\text{B}/\text{Br}$  ratio calculated for the thermal samples (i.e. 78.81 and 86.10  $\mu\text{g}/\text{L}$  for #1 and #2, respectively) results higher than the one characterizing seawater (#10: 48.83  $\mu\text{g}/\text{L}$ ) and freshwater (#5: = 0.08  $\mu\text{g}/\text{L}$ ). Differently, the  $\text{Cl}/\text{Br}$  ratio for thermal waters is equal to 211.71–198.12 mg/L resulting so lower than seawater value (248.10 mg/L) and comparable to the one from freshwater (203.45 mg/L). The thermal waters display also the highest computed  $\text{Sr}/(\text{Ca} + \text{Mg})$  and  $\text{Li}/\text{Cl}$  ratios (Fig. 12).

As shown in Tables 1–3 and Figs. 10–12, the chemical content of the other water samples varies from the thermal and sea waters composition (samples from the thermal springs and wells #3 and 4) or from the groundwater and seawater composition (cold springs and samples #6–#9). In particular, the seawater sample and the fresh one represent the end members of the mixing line plotted in the Langelier-Ludwig diagram (Fig. 10a) along which the cold-water samples (#6–#9 in Fig. 10a) are dispersed being them a

mixture of fresh groundwater and seawater. The mixing processes among fresh, thermal and seawater taking place at SCT system is peculiarly indicated by the waters sampled at different depths in well 4 (Fig. 10b). Here the shallower one (sample 4–125) indicate a composition of the fresh water fairly mixed with seawater while the deepest sample (sample 4–300) accounts for thermal water, similarly to #1 (Fig. 10b). A higher degree of mixing with seawater is recorded by #4–151 (Fig. 10b). In summary a geochemical water stratification is indicated as it should be expected.

The composition of the stripped gas obtained by the #2 thermal well indicated a content of about 72%  $\text{N}_2$ , 25%  $\text{CO}_2$  and of about 1%  $\text{CH}_4$ . These values suggest a relevant atmospheric  $\text{N}_2$ -rich component and a reduced component of deep  $\text{CO}_2$ . In fact, this  $\text{CO}_2$  having a  $\delta^{13}\text{C}$  of carbon at  $-1.95\text{\textperthousand}$  (PDB) clearly derives from the limestone dissolution. The corrected  ${}^3\text{He}/{}^4\text{He}$  ratio (Craig et al., 1978) was also measured in doubled. In terms of  $R/\text{Ra}$ , where  $R$  is the ratio in the sample and  $\text{Ra}$  the ratio in the air, the average value is of about 0.4 suggesting a crustal origin eventually affected by air ( $R/\text{Ra} = 1$  in air).

Regarding the isotopic composition of analysed samples, Fig. 13a shows the  $\delta\text{D}$ – $\delta^{18}\text{O}$  values compared with the Global Meteoric Water Line (GMWL; Craig, 1961) and the Mediterranean Meteoric

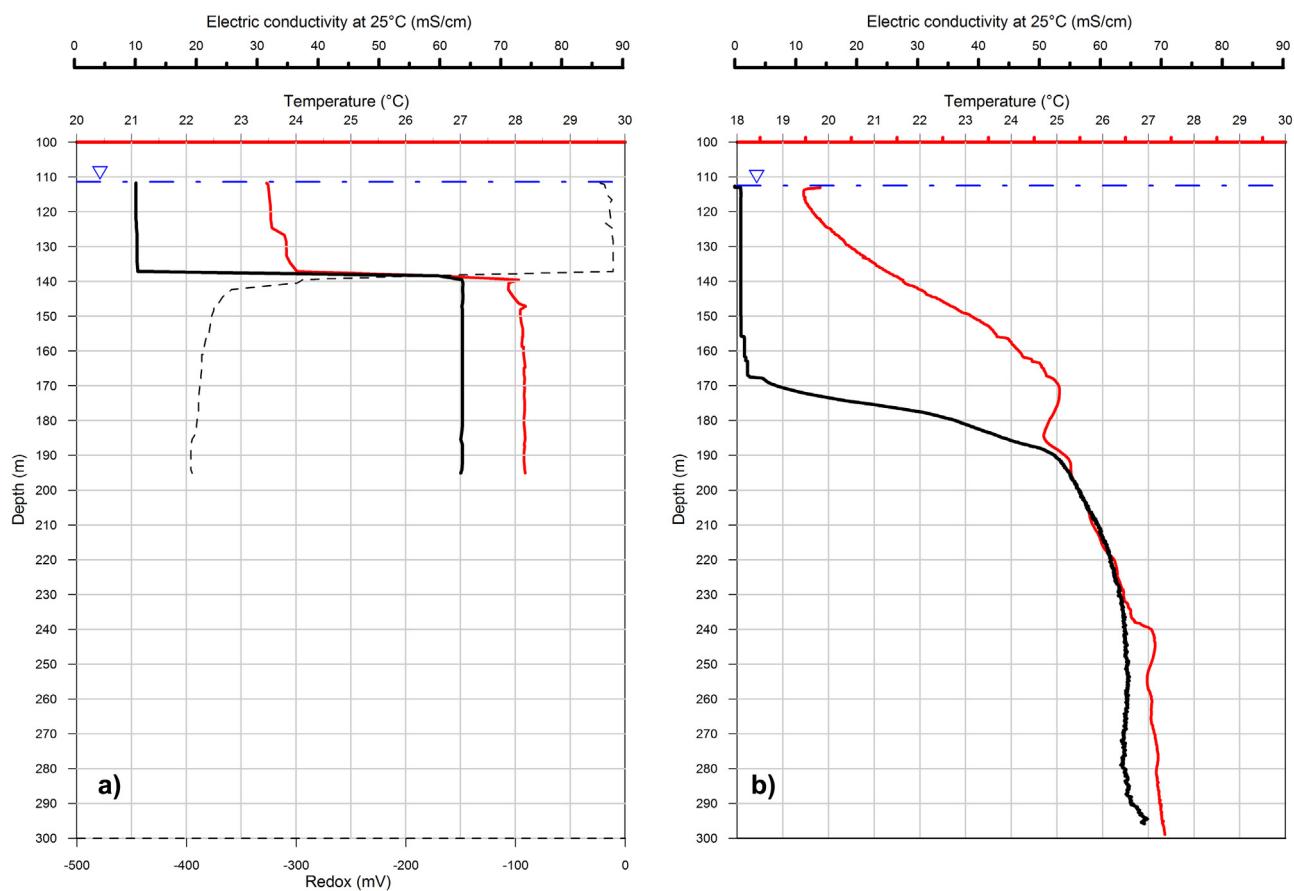


Fig. 9. Multiparametric logs for thermal wells 1 (a) and 4 (b).

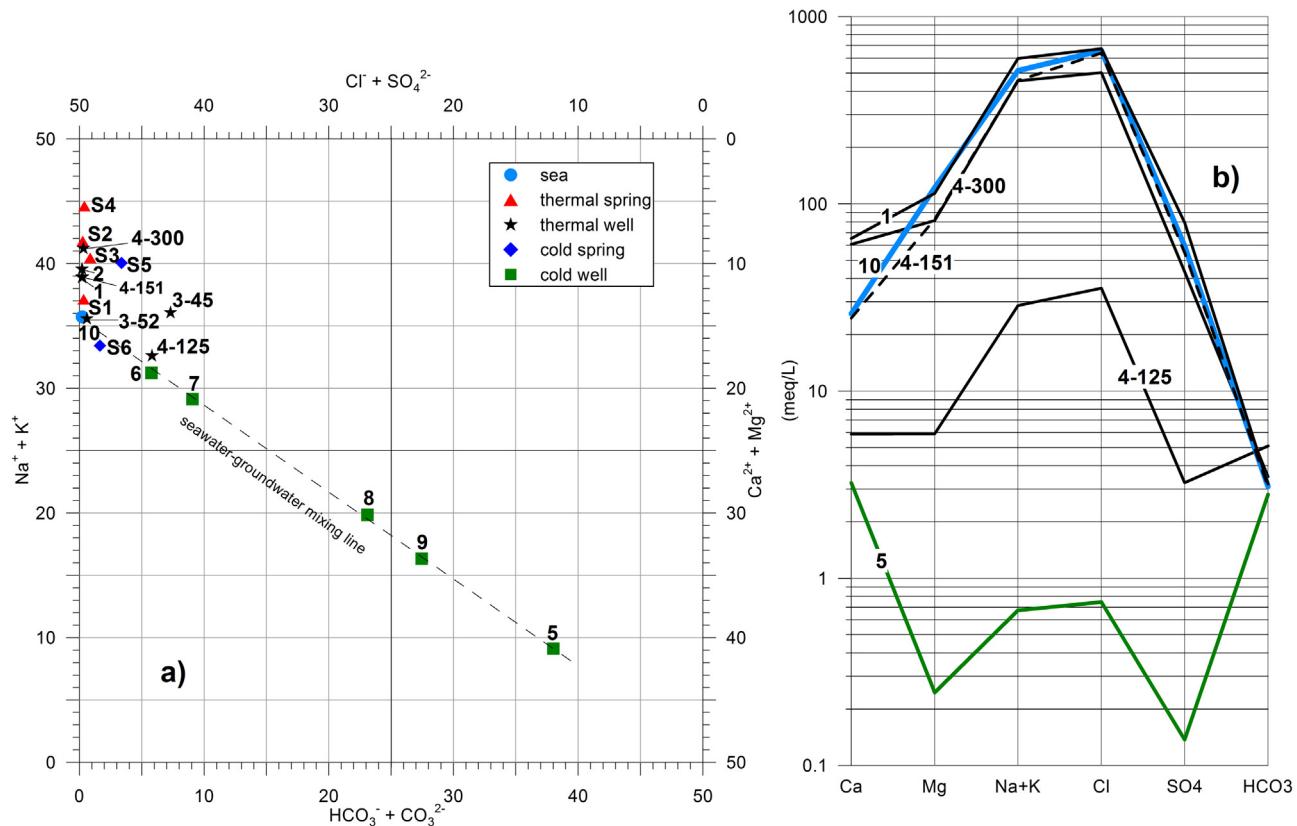
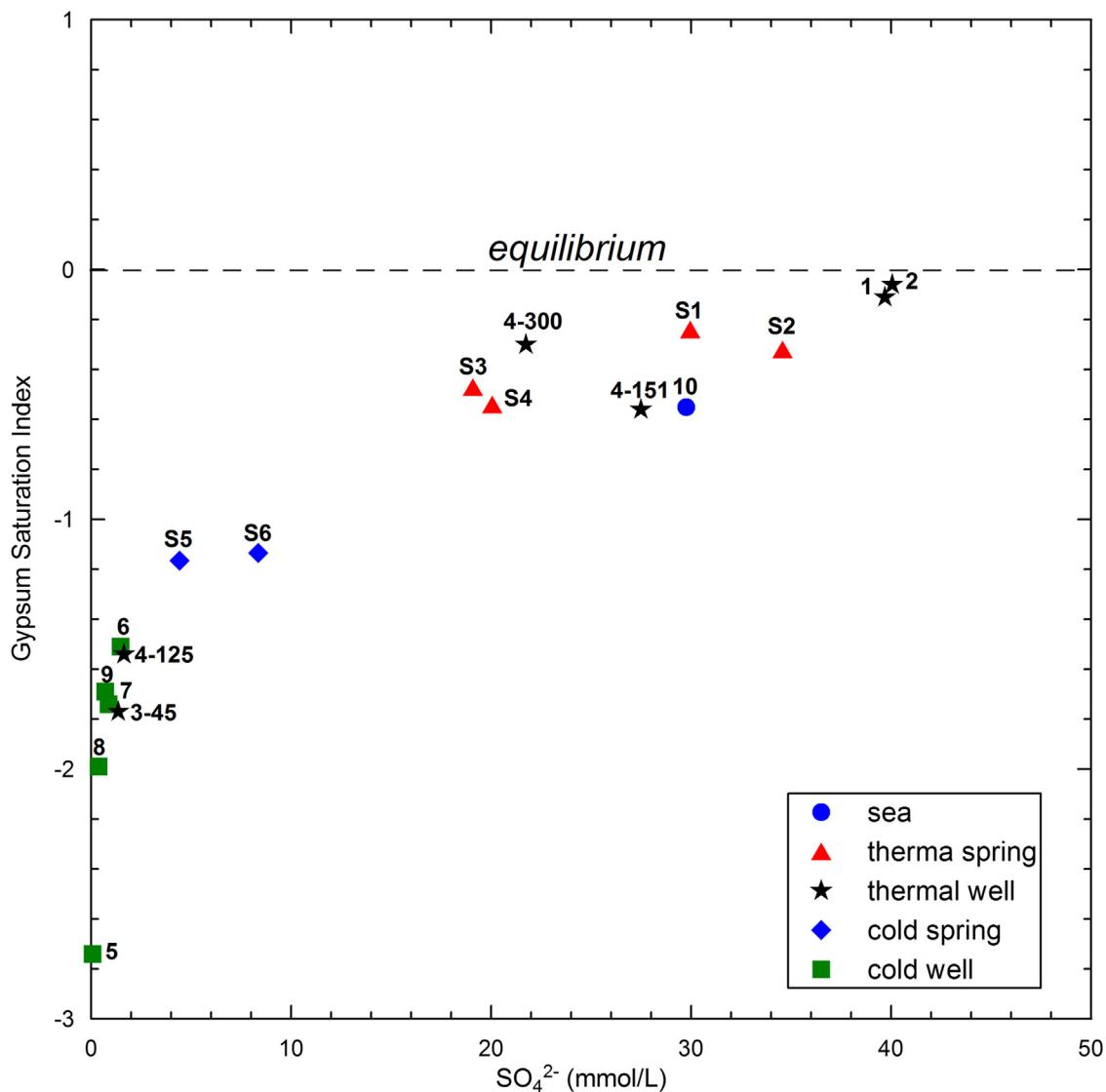


Fig. 10. Langelier-Ludwig (a) and Schoeller (b) diagrams of the sampled waters; data from Zuffianò et al. (2013) and new data.



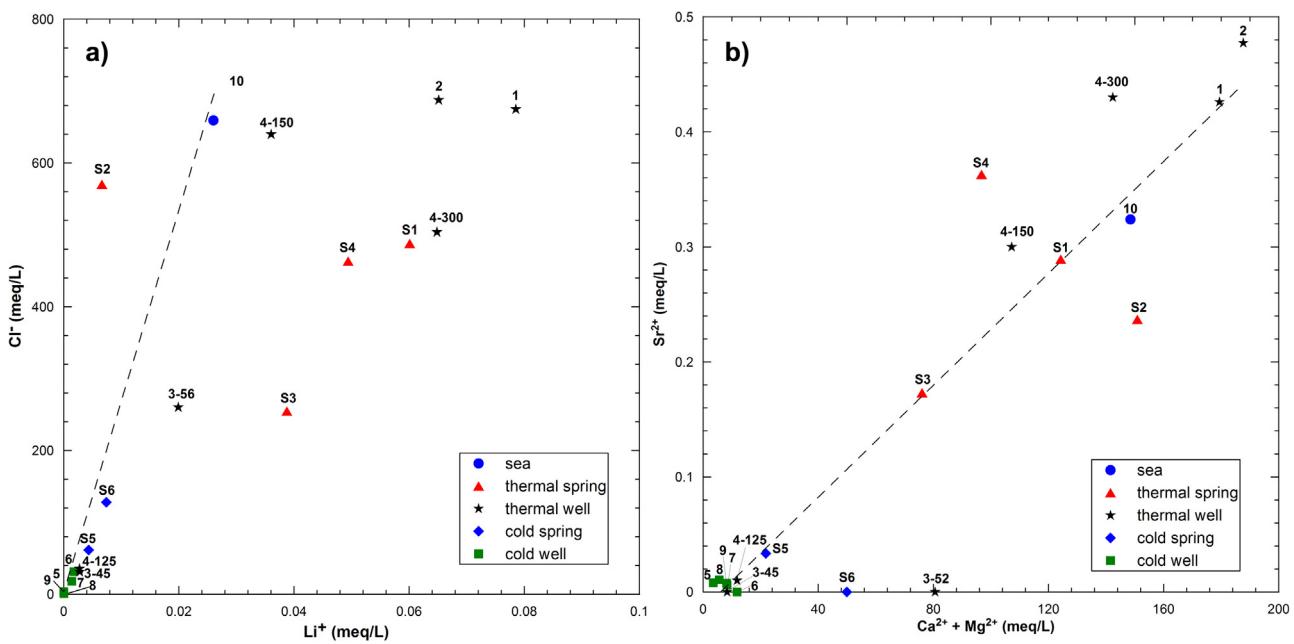
**Fig. 11.** Correlation between gypsum saturation index and sulphate concentrations in the sampled waters.

Water Line (MMWL; Gatt and Carmi, 1970). The Local Water Line (LWL), obtained interpolating all the analysed data, is also plotted in the same diagram. As a result, the cold water samples (#5–#9) fall between the MMWL and GMWL lines, thus suggesting a meteoric origin, with recharge areas located at altitudes less than 500 m a.s.l. (Minissale and Vaselli, 2011). All the other samples plot to the right of both these water lines (Fig. 13a), especially thermal samples #1 and #2 which record a slightly positive  $\delta^{18}\text{O}$ -shift with respect to seawater (Fig. 13a), probably connected to an evaporation process as here explained later on. The thermal spring waters (#S1, #S2, #S3, #S4) exhibit  $\delta\text{D}$  and  $\delta^{18}\text{O}$  values along a line between the meteoric samples (#5 to #9) and the saline ones, suggesting therefore the occurrence of a mixing of fresh, thermal and marine waters.

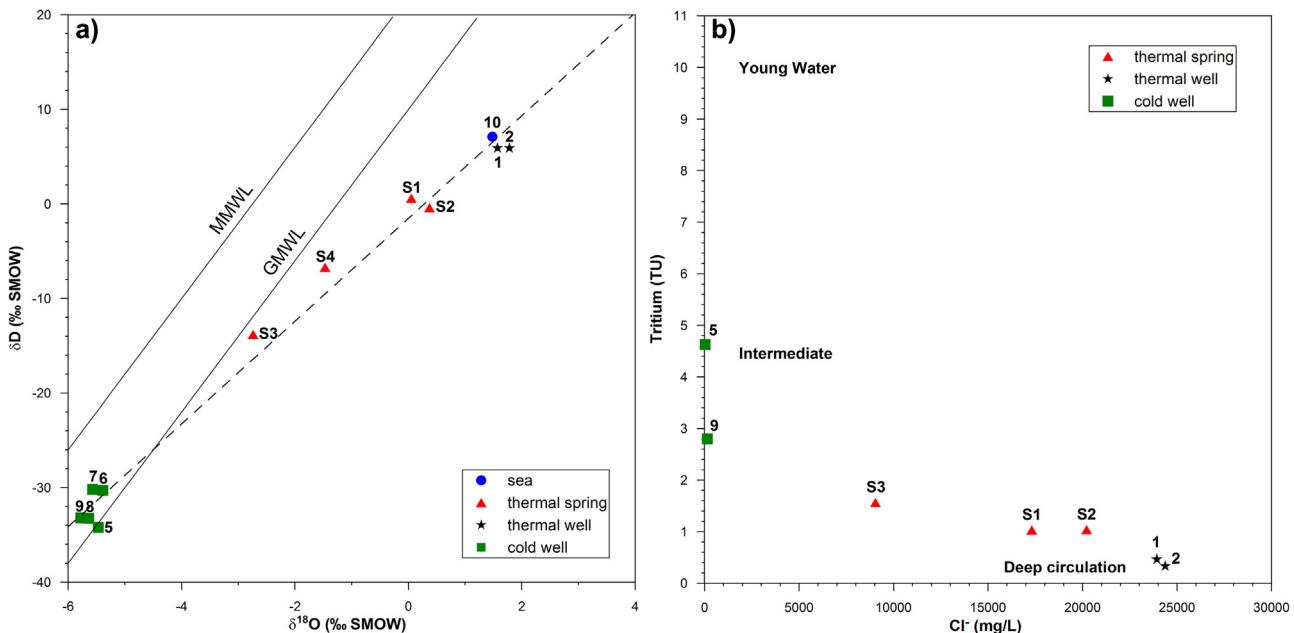
The thermal waters #1 and #2 have very low tritium content (0.46 and  $0.33 \pm 0.3$  TU), which is less than the present seawater content (2.2 TU; Sivan et al., 2005). Differently higher tritium contents, up to almost 5 TU, are recorded in the cold-water samples (Fig. 13b). These results suggest a very long residence time ( $>50$  years) of the thermal waters (pre-modern groundwater according to Clark et al., 1997)

Due to the large uncertainties affecting the chemical and physical processes (Plummer and Glynn, 2013), no carbon correction model was applied. Consequently, the water samples are considered as older than the uncorrected apparent ages. Therefore, the radiocarbon activities of the thermal waters ranges from  $7.54 \pm 0.8$  to  $8.45 \pm 0.8$  pmc corresponding to uncorrected  $^{14}\text{C}$  ages varying from 19849 to 21050 BP (Table 1), implying that recharge may have occurred during late Pleistocene. On the other hand, the  $^{14}\text{C}$  content obtained from a fresh water sample collected at well #6, far approximately 1 km north of thermal wells #1 and #2 (Fig. 2), provided an uncorrected age equal to 4268 BP, resulting very younger than the thermal ones.

In terms of sulphur isotopes, Fig. 14 shows the plot of  $\delta^{34}\text{S}$  and  $\delta^{18}\text{O}$  values, as collected in the sulphates from the thermal well waters (Table 4). These data are also displayed together with the ranges of the same isotopic ratios from groundwater samples, collected in different environments (Claypool et al., 1980; Utrilla et al., 1992; Clark and Fritz, 1997; Gunn et al., 2006). This plot indicates that the SCT sulphates are compatible with an interaction with Miocene evaporitic sediments that, in our case, can be related to the Messinian succession. Moreover, the isotope enrichment factor



**Fig. 12.** Plot of  $\text{Li}^+$  versus  $\text{Cl}^-$  (a) and of  $\text{Ca}^{2+} + \text{Mg}^{2+}$  versus  $\text{Sr}^{2+}$  (b) concentrations in all sampled waters; data from Zuffianò et al. (2013) and new data.



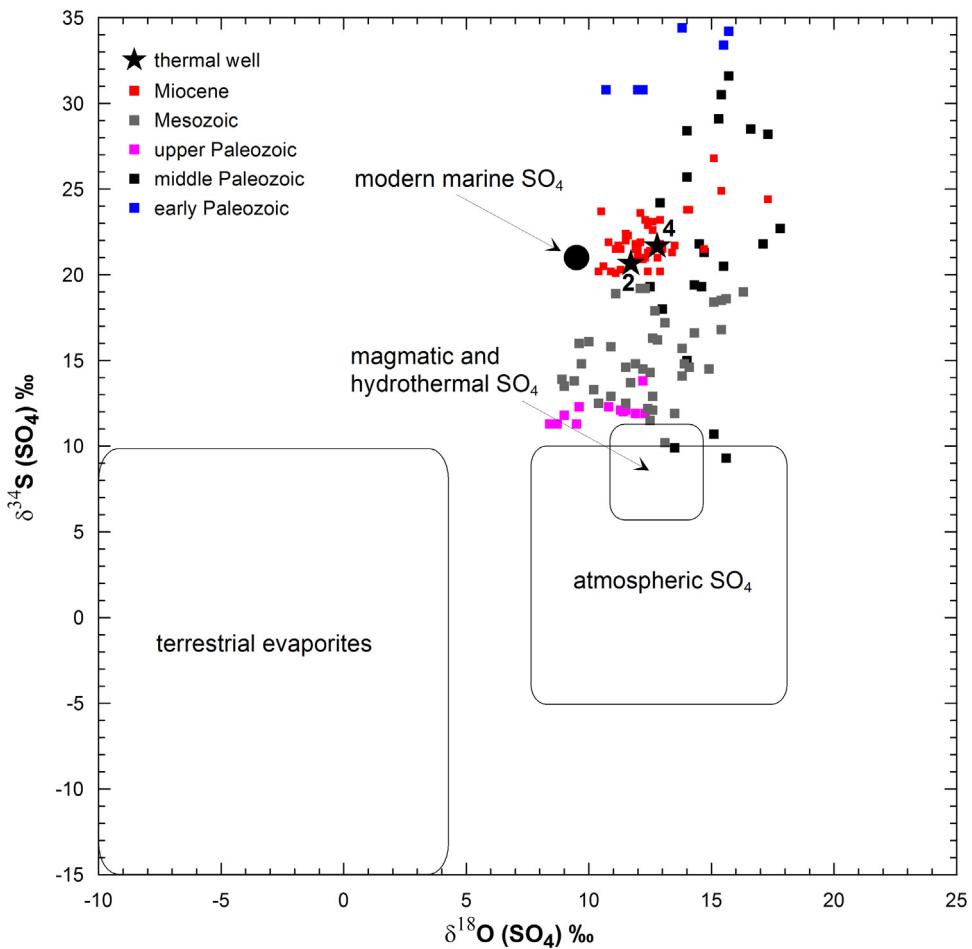
**Fig. 13.** (a) Binary  $\delta\text{D}-\delta^{18}\text{O}$  diagram for water samples; (b) Tritium-Cl correlation plot for sampled waters; data from Zuffianò et al. (2013) and new data.

**Table 4**  
Isotope values of groundwater sampled at the bottom of well 2 and 4.

| Sample | $\delta^{18}\text{O}$ |        | $\delta^{34}\text{S}$ |        | $\delta^{34}\text{S}$ |        |
|--------|-----------------------|--------|-----------------------|--------|-----------------------|--------|
|        | Result                | Repeat | Result                | Repeat | Result                | Repeat |
|        | VSMOW                 | VCDT   |                       |        |                       |        |
| 2      | 12.79                 | 12.32  | 21.67                 | 21.71  | -25.82                | -26.03 |
| 4      | 11.72                 | 11.81  | 20.66                 | 20.38  | -25.46                | -25.36 |

$\varepsilon$ , equal to 46.12‰, indicates a further bacterial sulphate reduction of dissolved sulphate (BSR; Wynn et al., 2010 and references therein).

Finally, the bulk of the thermal water samples plots in the field of partially equilibrated water, close to the immature waters field, within the Na-K-Mg $^{1/2}$  triangular diagram (Fig. 15), as proposed by Gigganbach (1988). This result implies a possible reservoir temperature encompassed between 160 and 180 °C. Only sample S3 indicates an immature water origin. Anyway, all the samples are far from the full equilibrium line, implying the geothermal water temperatures changed to a much greater extent during their re-equilibration processes. Other reliable geothermometers indicate temperatures between 25 and 28 °C and 25–37 °C for the Mg-Li and Na-Li geothermometers (Kharaka and Mariner, 1989) respectively. Differently, a temperature of 44–49 °C is estimated by the quartz geothermometer (Fournier, 1977).



**Fig. 14.** Ranges of  $\delta^{34}\text{S}$  and  $\delta^{18}\text{O}$  values of sulphates of various origins dissolved in groundwater compared with thermal well water samples (literature data are after Claypool et al., 1980; Utrilla et al., 1992; Clark and Fritz, 1997; Palmer et al., 2004; Gunn et al., 2006).

## 5. Discussion

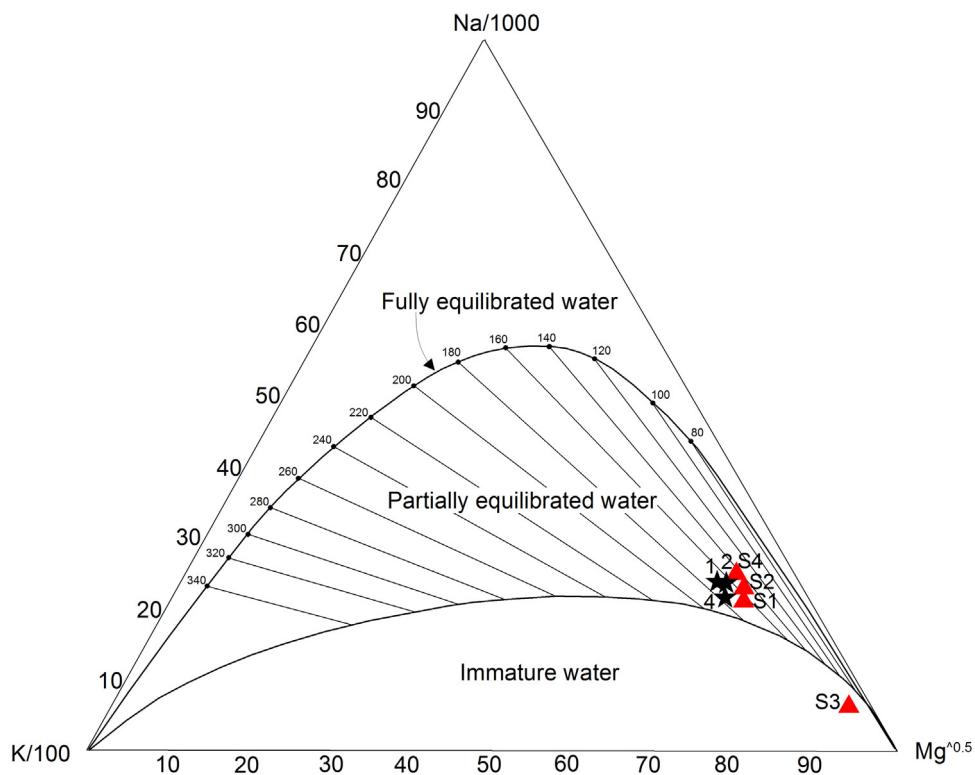
According to Calò and Tinelli (1995), the SCT thermal waters derive from upwelling seawaters through a system of closely spaced karstified tectonic fractures. Besides, the sulphate reduction of the seawater is to be related to reduction processes involving the organic matter bearing the calcareous rocks (Calò and Tinelli, 1995). Alternatively, organic substances and lignite content interlayered to Miocene outcrops can also interact with seawater giving rise to the same reduction processes (Zezza, 1980). Finally, a conrete origin of the thermal waters was also envisaged (Maggiore and Pagliarulo, 2004).

Some remarks from the previous authors are partly confirmed by the present research (e.g. seawater origin and sulphate reduction) although implemented by new data and considerations. Firstly, all the obtained hydro-geochemical data, delimit the area where the geothermal anomaly occurs: this is 1 km along the coastline and 2 km wide inland (Fig. 7), corresponding to the most faulted portion of the SCT zone (Fig. 2), affected by Holocene, almost vertical, NW-SE transtensional faults. Apart from fractures, as seen in detailed geological survey, the area is characterized by extensional jogs, indicating localized higher permeability, possible improved by karst processes. In this view, the hydrothermal minerals in the veins (Figs. 4–5) testify to the structural influence on channelling fluids. A similar set of faults was recognized offshore of the Santa Cesarea Geothermal area (Del Ben et al., 2015): notably some of these structures are considered seismically active (INGV, 2006).

Water geochemistry indicates three main types of waters: (a) the pure fresh groundwater (sample 5 in Figs. 10–13a) deriving from meteoric infiltration and flow in the limestone aquifer without mixing with seawater intrusion; (b) brackish waters, from the partial mixing between fresh groundwater and saline groundwater due to seawater intrusion (e.g. samples 9, 8, 7, 6, 4–125, 4–151 in Figs. 10–13) produced by different amounts of seawater content (the highest in sample 4–151, Fig. 10); (c) thermal groundwater, occurring at depth (samples 1 and 2; Fig. 9). Thermal groundwater interacts with both fresh groundwater and saline groundwater of seawater intrusion, giving rise to variable chemical compositions, as it is the case from samples from Fetida and Gattulla caves (samples #S1 and #S3). For this reason, the geochemical features of the geothermal reservoir are defined by the chemical composition from wells #1 and #2 and #4 at depth (Fig. 10).

The chemical composition of the thermal waters can provide information on the fluid pathway. In fact, with respect to fresh groundwater and seawater, the higher  $\text{Ca}^{2+}$  and low  $\text{Mg}^{2+}$  concentrations could derive from calcite dolomitization (Collins, 1975), as commonly it occurs in limestone. On the other hand, the increase of  $\text{Ca}^{2+}$  content could be related to the dissolution of gypsum or anhydrite, generally included in the Messinian evaporite where the concomitant bacterial  $\text{SO}_4^{2-}$  reduction can develop (Appelo and Postma, 1996).

Furthermore, the significant amount of Br and B contents could be the result of both organic matter decomposition (Martin et al., 1993; You et al., 1993) and water interaction with illite/smectite



**Fig. 15.** Na/1000-K/100-Mg<sup>1/2</sup> ternary diagram (Giggenbach, 1988) applied to thermal waters.

clays (Kharaka and Hanor, 2003). Besides, boron tends to be concentrated in environments with a limited water circulation, such as the case of evaporite and brines (Uhlman, 1991). More other information on the fluid path is taken from the high Li<sup>+</sup> content of which amount indicates interaction with clay minerals (White, 1957; Paropkpari, 1990) or a long residence in reservoirs (Edmunds and Smedley, 2000). Even the high content of As estimated for the thermal waters respect to the other samples could be related to the water interaction with evaporitic minerals or rocks including limestone/dolomite (Dogan and Dogan, 2007).

The overall positive δ<sup>18</sup>O and δD shift recorded by samples #1 and #2 should reflect the composition of an original seawater undergone evaporation effects (Boschetti et al., 2011; Clark and Fritz, 1997).

The geothermometers give a range of the temperature reservoir between 25 °C and 160 °C. Accepting the temperature upper limit of 80–110 °C given by the bacterial sulphate reduction (Machel, 2001; Wynn et al., 2010), this range narrows.

During the upwelling towards the springs' outflows, the deep thermal water, the lesser dense water in the SCT system due to the highest temperature, notwithstanding the highest salinity, breaks the barrier defined by the occurrence of both the intruded seawater (the more dense water) and the cold groundwater (denser than the thermal one but less dense than the intruded seawater). The higher piezometric levels recorded by the thermal wells (e.g. in well 4: 1 m a.s.l.) respect to the cold groundwater confirm in fact an upward flowing.

In this view, with the knowledge that the water density is a temperature and salinity function, and being the last parameter almost constant at depth (Fig. 7), thermal water temperature should be high enough to favour its upflowing. Based on the equation from McCutcheon et al. (1993), a temperature of 80 °C has been estimated as minimum necessary to promote the thermal water upflowing. Finally, based on the equation from Liu et al. (2015), the

estimated circulation depth of 2200 m has been calculated considering, in the study area, the average temperature gradient of about 3 °C/100 m (Mongelli et al., 1983), a mean annual air temperature of about 18 °C and a depth of the constant-temperature zone of 30 m (Romanazzi et al., 2015).

## 6. Conclusion

Fig. 16 summarizes the hydrothermal circulation pathway for the SCT system as resulted from the integration of data, as previously described.

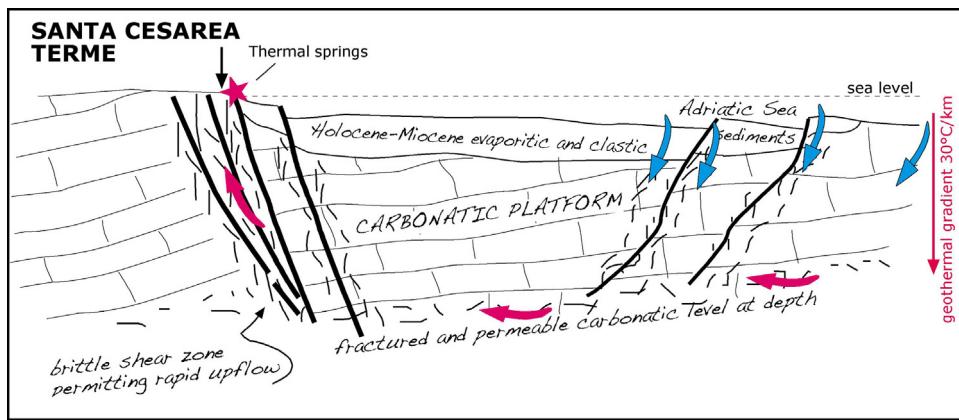
The recharge area is located offshore, at sea bottom, and, in this context, it is channelled through localized fault damages zones. Faults, affecting both the Miocene-Quaternary sediments and the underlying carbonate substratum, are supposed with a lateral to oblique component as indicated by focal mechanism on recent seismic events.

Moving downwards along the faulted zones, infiltrating seawater is heated (due to the geothermal gradient) and interacts with rocks of different origin (i.e. Messinian evaporates and Mesozoic-Oligocene limestone, mainly). The result is thus an enrichment, for instance, in calcium, sulphate, boron, sulphide and lithium elements, making the original seawater a thermal fluid, as indicated by oxygen isotope ratios, as well.

Thereafter, the geothermal waters, already heated by the geothermal gradient, are driven upwards by the buoyant forces generated by the density decrease. Along the almost vertical structural channels (e.g. extensional jogs) determined within the transtensional damage fault zones.

The circulation path here proposed here lasts a long time, roughly 20000 years as indicated by the geochronological analyses.

In conclusion, the SCT springs are evidence of a hydrothermal-convection system, where convection occurs within the fault damage zones (Fig. 16), passing through the Messinian deposits and



**Fig. 16.** Conceptual model (not to scale) of the relationships between brittle structures and fluid flow in the Santa Cesarea Terme area. The seawater is channelled (blue arrows) to depth through the damage zones of the faults affecting the sea-bottom down to the carbonate platform at depth, interacting mainly with Messinian evaporites and Mesozoic-Oligocene limestone. At a deeper structural level, a fractured and permeable level is supposed to host the marine water that has been infiltrated through the carbonate platform. Conversely, the fault system in Santa Cesarea determined a structural channel where the marine water, can rapidly upflows, maintaining a part of the temperature gained at depth. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

the Mesozoic-Oligocene carbonate rocks, capped by the Holocene-Pliocene sediments. The original seawater therefore mostly resides at a depth of about 2–3 km, where it reaches a temperature of about 85 °C, from where it moves upwards, crossing the transition zone of the seawater intrusion.

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